



WBCS 2022



SCIENCE

POLLUTION PART-2



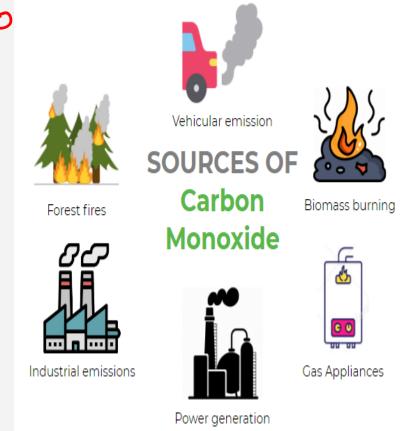




Oxides of Carbon (i) Carbon monoxide:

- ☐ Carbon monoxide (CO) is one of the most serious air pollutants.
- ☐ It is produced as a result of incomplete combustion of carbon.

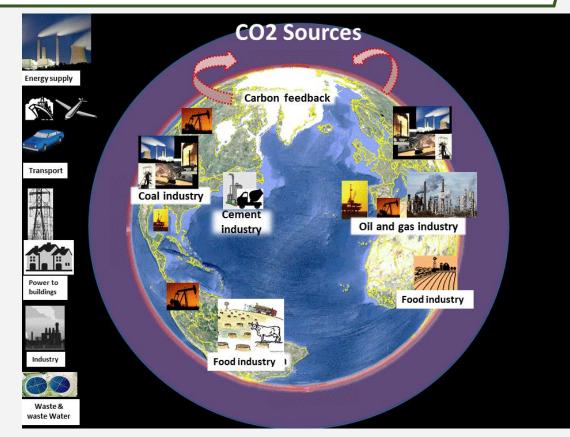
 Carbon monoxide is mainly released into the air by automobile exhaust.
- Other sources, which produce CO, involve incomplete combustion of coal, firewood, petrol, etc.
- It binds to haemoglobin to form carboxyhaemoglobin, which is about 300 times more stable than the oxygen-haemoglobin complex.
- □ In blood, when the concentration of carboxyhaemoglobin reaches about 3–4 per cent, the oxygen carrying capacity of blood is greatly reduced. This oxygen deficiency, results into headache, weak eyesight, nervousness and cardiovascular disorder.

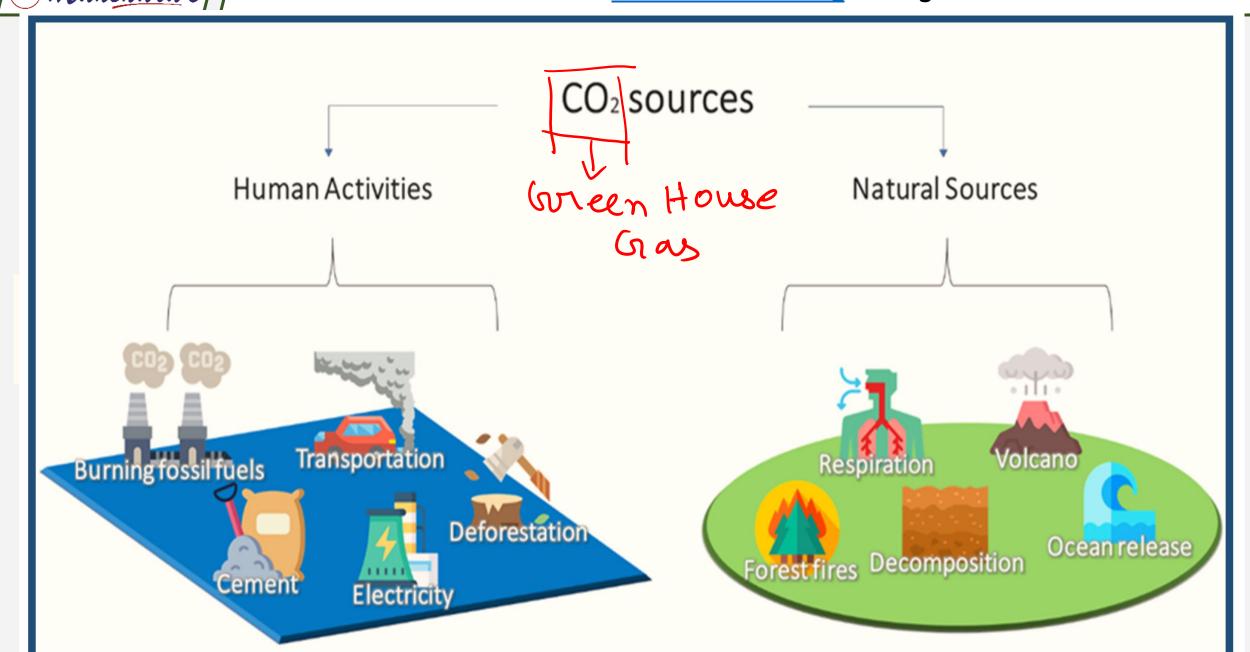




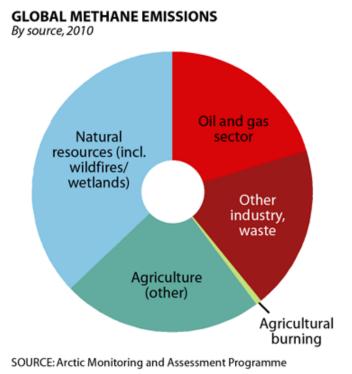
Carbon dioxide:

- Carbon dioxide (CO2) is released into the atmosphere by respiration, burning of fossil fuels for energy, and by decomposition of limestone during the manufacture of cement.
- 2. With the increased use of fossil fuels, a large amount of carbon dioxide gets released into the atmosphere.
- Excess of CO2 in the air is removed by green plants and this maintains an appropriate level of CO2 in the atmosphere.
- 4. Green plants require CO2 for photosynthesis and they, in turn, emit oxygen, thus maintaining the delicate balance.

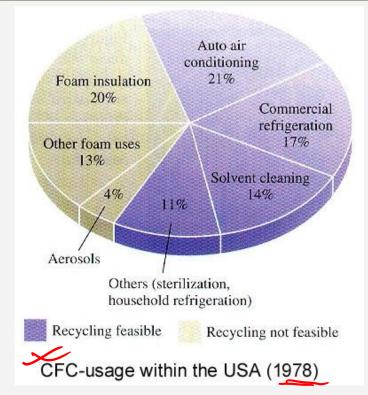








1. Methane is produced naturally when vegetation is burnt, digested or rotted in the absence of oxygen.



1. Chlorofluorocarbons (CFCs) are man-made industrial chemicals used in air conditioning etc. CFCs are also damaging the ozone layer.



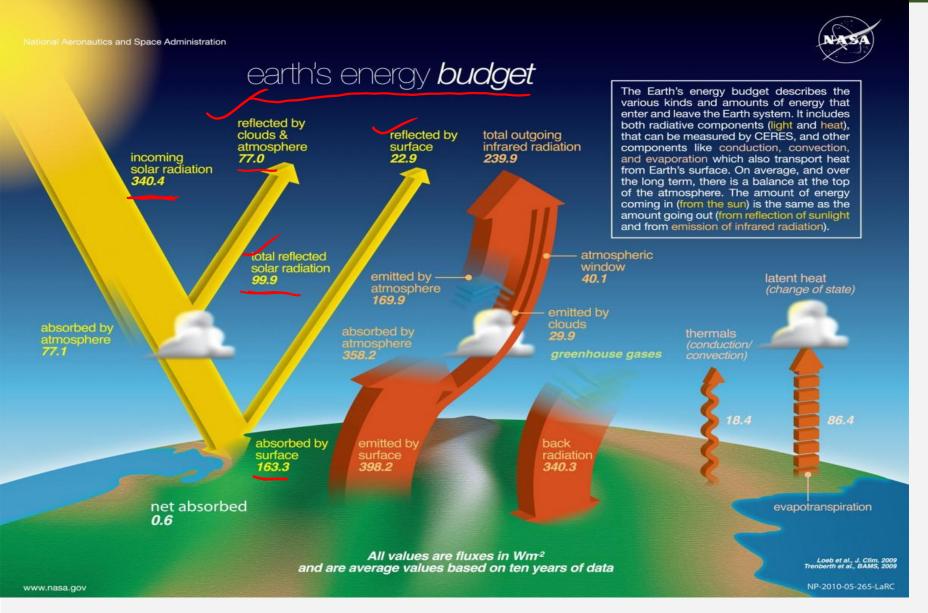
Global Warming and Greenhouse Effect

- 1. About 75 % of the solar energy reaching the earth is absorbed by the earth's surface, which increases its temperature.
- 2. The rest of the heat radiates back to the atmosphere. Some of the heat is trapped by gases such as carbon dioxide, methane, ozone, chlorofluorocarbon compounds (CFCs) and water vapour in the atmosphere.
- 3. Thus, they add to the heating of the atmosphere. This causes global warming.
- 4. Besides carbon dioxide, other greenhouse gases are methane, water vapour, nitrous oxide, CFCs and ozone.



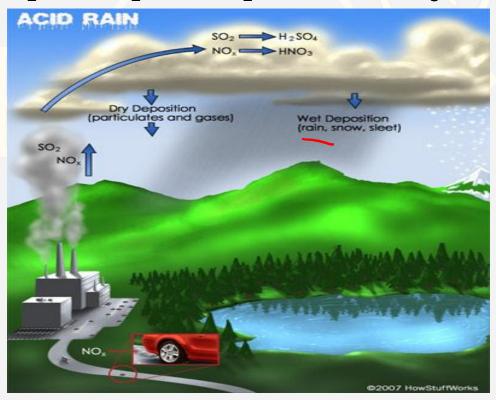


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Acid rain:

- When the pH of the rain water drops below 5.6, it is called acid rain.
- $2SO_2(g) + O_2(g) + 2H_2O(I) \rightarrow 2H_2SO_4(aq)$
- $4NO_2(g) + O_2(g) + 2H_2O(I) \rightarrow 4HNO_3(aq)$



- Acid rain is a by product of a variety of human activities that emit the oxides of sulphur and nitrogen in the atmosphere.
- ❖ SO2 and NO2 after oxidation and reaction with water are major contributors to acid rain, because polluted air usually contains particulate matter that catalyse the oxidation.



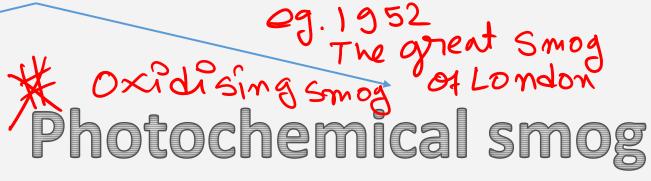
SMOG

The word smog is derived from smoke and fog

Fog + Soz + Dust

CLASSICAL SMOG



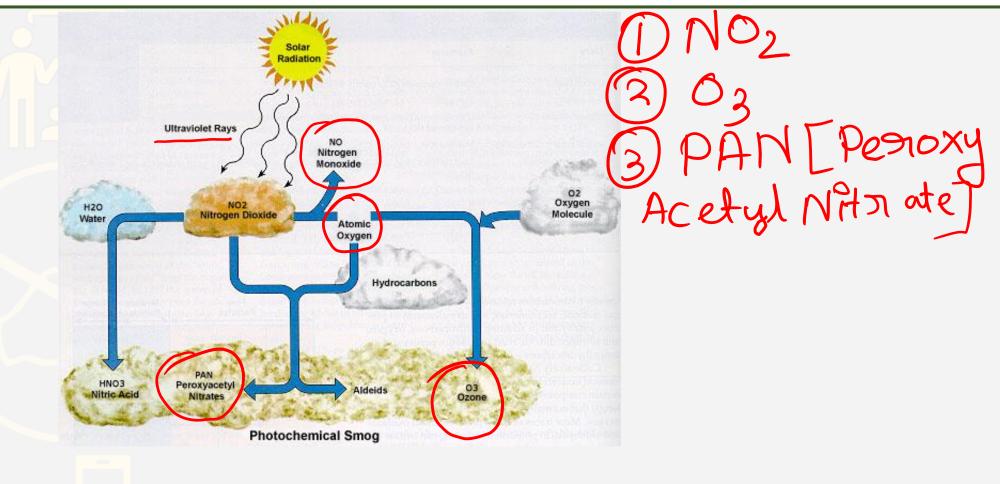






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Stratospheric Pollution Formation and Breakdown of Ozone

The upper stratosphere consists of considerable amount of ozone (O_3) , which protects us from the harmful ultraviolet (UV) radiations (λ 255 nm) coming from the sun. These radiations cause skin cancer (melanoma) in humans. Therefore, it is important to maintain the ozone shield.

$$\begin{split} & O_{2(g)} \xrightarrow{UV} O_{(g)} + O_{(g)} \\ & O_{2(g)} + O_{(g)} \xleftarrow{UV} O_{3(g)} \\ & CF_2Cl_{2(g)} \xrightarrow{UV} \dot{C}l_{(g)} + \dot{C}F_2Cl_{(g)} \\ & \dot{C}l_{(g)} + O_{3(g)} \longrightarrow Cl\dot{O}_{(g)} + O_{2(g)} \\ & \dot{C}l_{(g)} + O_{(g)} \longrightarrow \dot{C}l_{(g)} + O_{2(g)} \end{split}$$



WBCS BATCH 2022



















(GA/GS)

LIVE 11:30 AM ONWARDS

(ENGLISH)



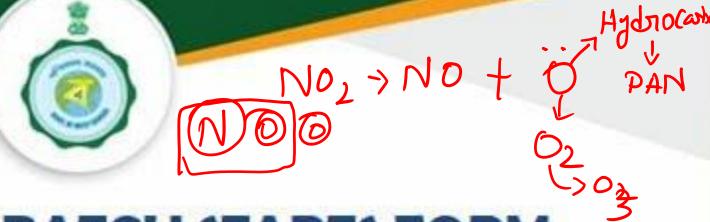
W.B.C.S -YouTube schedule for 5-days

TIMING	DAYS				
	MONDAY	TUESDAY	WEDNESDAY	THURSDAY	FRIDAY
11:30	ECO	CA	ECO	POL	BEN
12:00	REAS	SCI	POL	SCI	CA
12:30	ENG	MATHS	ENG	MATHS	ENG
13:00	GEO	HIS	GEO	HIS	HIS



WBCS BATCH 2022

 $0+0 \Rightarrow 0_{2}$ $0+0 \Rightarrow 0_{6} H_{12} O_{6} + 0_{2}$ Photosymphesis



WBCS OFFLINE BATCH STARTS FORM

26TH FEBRUARY 5th March

TIMING - 1:30-5:30PM

BOOK YOUR SEAT NOW

9230141497/8017652045/6386903177/6291322109

